

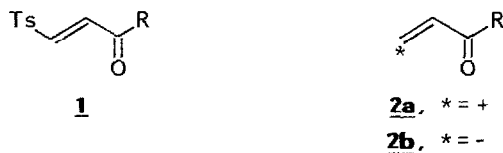
STEREOSELECTIVE SYNTHESIS OF (E)- β -ALKYL ACRYLATES AND ACRYLAMIDES

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Summary: β -Tosyl- α,β -unsaturated esters and amides react regio and stereoselectively with carbanionic reagents to afford β -alkyl, alkenyl, or alkynyl substituted α,β -enoates or enamides with an E configuration.

β -Sulfonyl- α,β -unsaturated carbonyl compounds **1**¹ are interesting molecules because they exhibit dual reactivity at their β -position: they can act either as β -acylvinyl cation equivalents **2a**¹ or as β -acylvinyl anion ones **2b**², according to a normal or umpoled reactivity, respectively. In both cases the sulfone group directs the attack at the β -position either by nucleophilic or electrophilic reagents, when the last process was preceded by the corresponding metallation reaction. The reactivity of compounds of the type **1** as β -acylvinyl cation equivalents involves the substitution of the sulfone moiety by a nucleophile and has been mainly described for β -sulfonylvinyl ketones and nitrogen or sulfur containing nucleophiles;^{1,3} this process is facilitated by other electron-withdrawing groups such as nitro⁴ or sulfone.⁵ We now report the synthesis of β -alkyl substituted α,β -enoates⁶ and enamides from β -tosylacrylic derivatives and carbanionic reagents, the sulfone group acting as a leaving one.



The reaction of the (E)- β -tosyl- α,β -unsaturated esters **3a,b** and (E)-N-(β -tosylacryloyl)piperidine (**3c**) with two equivalents of the corresponding organomagnesium compound⁷ at 0°C in tetrahydrofuran led regio and stereoselectively to (E)- α,β -unsaturated esters or amides **4** (Scheme 1 and Table 1). When vinylmagnesium bromide was used as nucleophile the

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Table 1 . Reaction of β -tosylated acrylic derivatives **3** with organometallic reagents. Preparation of β -alkyl- α,β -unsaturated esters and amides **4** or compounds **5**.

Starting material			Organometallic	Reaction	Product ^a		
No.	R ¹	X	R ² M	time (h)	No.	Yield (%) ^b	B.p. (°C) ^c or R _f ^d
3a	H	OEt	CH ₂ =CHMgBr	3	4a	79	80
3c	H	$\overline{\text{N}(\text{CH}_2)_5}$	CH ₂ =CHMgBr	2	4b	90	160
3c	H	$\overline{\text{N}(\text{CH}_2)_5}$	<i>i</i> -PrMgBr	14	4c	56	0.25
3a	H	OEt	<i>n</i> -BuMgBr	1	4d	75	60
3b	Me	OMe	<i>n</i> -BuMgBr	4	4e	89	60
3a	H	OEt	<i>n</i> -C ₆ H ₁₃ C≡CMgBr ^e	24	4f	55	0.75
3b	Me	OMe	(EtO ₂ C) ₂ CHNa ^f	15	4g	70	160
3a	H	OEt	(EtO ₂ C) ₂ CHNa ^f	1	5a	60	0.65 ^g
3c	H	$\overline{\text{N}(\text{CH}_2)_5}$	(EtO ₂ C) ₂ CHNa ^f	24	5b	78	0.46 ^g

^a All compounds are >95% pure (g.l.c.) and gave satisfactory spectral data (i.r., ¹H and ¹³C n.m.r., and mass spectra).

^b Based on the starting sulfone **3**. Yields after distillation or flash chromatography. Products **4** were isolated as pure (*E*)-stereoisomers (g.l.c. or/and 300 MHz ¹H n.m.r.).

^c Bath temperature at 0.1 torr.

^d Silica gel; hexane/ether: 1/2.

^e Prepared from 1-octyne and isopropylmagnesium bromide.

^f Prepared from diethylmalonate and sodium hydride.

^g Silica gel; hexane/ether: 1/10.

6. For synthesis of β -alkyl substituted α,β -enones and enoates by means of organocuprates and α,β -unsaturated carbonyl compounds containing a good leaving group at the β carbon atom see: R. K. Dieter and L. A. Silks III, J. Org. Chem., **1986**, 51, 4687, and references cited therein.
7. For transition metal catalyzed substitution of the sulfone group in vinyl sulfones by means of Grignard reagents see: J. L. Fabre, M. Julia, and J. N. Verpeaux, Tetrahedron Lett., **1982**, 23, 2469.
8. For preparation of dienoates and dienamides and their application in the synthesis of natural products see, for instance: (a) S. Tsuobi and A. Takeda, Tetrahedron Lett., **1979**, 1043; (b) T. Sakai, K. Seko, A. Tsuji, M. Utaka, and A. Takeda, J. Org. Chem., **1982**, 42, 1101; (c) B. M. Trost, M. Lautens, and B. Peterson, Tetrahedron Lett., **1983**, 24, 4525; (d) S. Tsuobi, Y. Nooda, and A. Takeda, J. Org. Chem., **1984**, 49, 1204; (e) T. Mandai, T. Moriyama, K. Tsujimoto, M. Kawada, and J. Otera, Tetrahedron Lett., **1986**, 27, 603; (f) R. Bloch, J. Abecassis, and D. Hassan, Can. J. Chem., **1984**, 62, 2019.
9. In this case sodium hydride was added to the solution of diethyl malonate in tetrahydrofuran.
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